

<i>Programme</i>	Chemistry BSc
<i>Course title</i>	Physical Chemistry (1)
<i>Name of educator</i>	Ernő Keszei, György Inzelt
<i>Type of course</i>	<u>compulsory</u> , semi-optional, elective
<i>Module</i>	non-chemical, <u>core-chemical</u> , specialized chemical, chemistry teacher
<i>Course code</i>	KA2FZ1 + KA2FZ2
<i>Number of credits</i>	3+1
<i>Year of study</i>	1
<i>Semester</i>	fall, <u>spring</u>
<i>Form of tuition</i>	<u>lectures</u> , <u>practice</u> , laboratory practice, other
<i>Course contents</i>	The course presupposes the knowledge acquired during the courses physics (1) and calculus. Based on this knowledge, the course begins with a postulatory development of the basics of phenomenological thermodynamics, with an outlook to the classical foundations of the topic. A more detailed discussion of a number of chemical applications follows, including mixtures, phase equilibria, chemical equilibria and the chemical thermodynamics of electrically charged phases. A brief discussion of the principles of general transport phenomena is followed by some details of diffusion, viscous flow and electric conduction. The focus in applied electrochemistry is on the equilibrium description of galvanic cells.
<i>Assessment method</i>	<u>written/oral examination</u> , practical course mark, other
<i>Recommended reading</i>	P.W. Atkins: Physical Chemistry, 7 <sup>th</sup> edition, Oxford, 1998 H.B. Callen: Thermodynamics, New York, 1985
<i>Language of instruction</i>	Hungarian